

CHEMISTRY



2022
HANDBOOK

 **DUT**
DURBAN UNIVERSITY OF TECHNOLOGY
INYUVESI YASETHEKWINI YEZOBUCHWEPHESHE

 **FACULTY OF
APPLIED
SCIENCES**

2022 HANDBOOK: CHEMISTRY

FACULTY of
APPLIED
SCIENCES

**DEPARTMENT of
CHEMISTRY**

IMPORTANT NOTICES

- Your registration is in accordance with all current rules of the Institution. If, for whatever reason, you do not register consecutively for every year/semester of your programme, your existing registration contract with the Institution will cease. Your re-registration anytime thereafter will be at the discretion of the institution and, if permitted, will be in accordance with the rules applicable at that time.
- The rules in this departmental handbook must be read in conjunction with the General Rules (G Rules) contained in the DUT General Handbook for Students as well as the relevant Study Guides.
- With respect to an appeal, your attention is specifically drawn to Rules GI (8) and (9), and to the process of dealing with students' issues.

STRATEGIC DIRECTION (2020-2030) FACULTY OF APPLIED SCIENCES

[Educate. Innovate. Engage]

VISION

Leading innovation through science and technology

MISSION STATEMENT

Educate students

Generate new scientific
knowledge Engage

communities

VALUES

1. **Accountability:** We take ownership of all activities, resources and tasks required of us. We deliver on our promises and responsibilities.
2. **Integrity:** We adhere to moral standards and principles. We are transparent and consistent in all our actions, and lead by example.
3. **Dedication:** We are committed to achieving our goals and expectations.
4. **Professionalism:** We operate within clear boundaries with respect to our code of conduct.
5. **People Oriented:** We are committed to sustaining the morale and holistic development of staff and student. We value diversity in all forms.

DEPARTMENT OF CHEMISTRY

MISSION

Advancing future chemists for industry and research

VISION

- Making chemistry fun through science, technology and innovation

VALUES

- Trust - (We always speak the truth. We are people of integrity)
- Respect - (We listen to the opinion of others. We treat others with dignity. Our individuality is valued. We subscribe to the spirit of Ubuntu)
- Communication - (We seek to understand others better. We follow through and meet deadlines. We eliminate confusion and promote clarity)
- Passion - (We love chemistry. Chemistry is fun. Our teaching and learning methods are exciting. We take pride in what we do)

Teamwork

(Our productivity and outputs reflect our team spirit. Our enthusiasm is infectious)

Accountability

(We do what we say. We are responsible and reliable. We take initiative. We give credit where credit is due. Our success is a result of what we do)

Excellence

(We lead, others follow. Whatever we do, we do it right first time. Quality matters. We pride ourselves in academic excellence and innovative research)

"We Are Chemistry"

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I. DEPARTMENTAL & FACULTY CONTACT DETAILS

All departmental queries to:

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Fax to Email No: 0866740608
Email: TBA
Location of Department: S10 Level 3, Steve Biko Campus

All Faculty queries to:

Faculty Officer: Ms G Shackleford
General Enquiries No: 031 373 2506
Facsimile No: 031 373 2175
Email: fas@dut.ac.za
Location: Block S4 Level 3, Steve Biko Campus

Faculty Assistant: Ms NP Ngwazi
General Enquiries No: 031 373 2717
Facsimile No: 031 373 2715
Email: nonhlanhlaN3@dut.ac.za
Location: Block S4 Level 3, Steve Biko Campus

Executive Dean:

Executive Dean's Secretary: Prof S Singh
Telephone No: Mrs NK Naidoo
Facsimile No: 031 373 2720
Email: 031 373 2724
Location: fas@dut.ac.za
Steve Biko Campus Between Block S6 and S7, Level 4

2. DEPARTMENTAL STAFF

Head of Department	Prof N Deenadayalu, BSc (Hons) (UN), MSc (UN), PhD (UN)
Professors	Prof K Bisetty, BSc (Hons) (UDW), UHDE (UDW) NHDip (MLST), MSc (UN), PhD (UN) Prof RM Gengan, BSc (Hons) (UDW), HED (UNISA), NHDip (MLST), MSc (UN), PhD (UN)
Associate Professors	Prof PS Mdluli, BSc (Hons) (UZ), MSc (UKZN), PhD (UZ)
Senior Lecturers	Dr TH Mokhothu, BSc (Hons) (UFS), MSc (UFS), PhD (UFS) Dr V Paul, MTech (MLST), BTech: BusAdmin (DUT), PhD (DUT) Dr MI Sabela, BTech (DUT), MTech (DUT), PhD (DUT)
Lecturers	Dr TR Makhanya, BTech (DUT), M Tech (DUT), PhD (DUT) Dr P Ntola, MTech (DUT), PhD (UKZN) Dr SS Ndlela, BSc (Hons) (UKZN), MSc (UKZN)
Junior Lecturer	Ms H Lubanyana, BSc (Hons) (UKZN), MSc (UKZN)

Secretary	Ms P Ntetha
Senior Technicians	Mrs CT Gwamanda, BTech (TUT) Mr N Ramnarayan, NDip (MLST)
Technicians	Ms NM Xhakaza, BTech (MLST), MTech(DUT) Ms D Naicker, BTech (MLST) Mr R Ramkrepal, BTech (MLST) Ms NP Cele, BTech (MLST), MTech (DUT) Ms SS Zunngu, BTech (DUT), MAppSci (DUT) Mr SL Majola, NDip (DUT), BTech (DUT) Mr ZI Miya, BSc (Hons) (UKZN) Mr S Sithole, NDip (DUT), BTech (DUT) Mr TN Cele, NDip (DUT), BTech (DUT)
Laboratory Assistants	Ms NN Mpungose, NDip (DUT), BTech (DUT)
Emeritus Professors	Prof KG Moodley Prof GG Redhi

3. QUALIFICATIONS OFFERED BY THE DEPARTMENT

Programmes offered in this Department, which, upon successful completion, will lead to the award of the following qualifications:

- Diploma in Analytical Chemistry (D)
- ND: Analytical Chemistry (ND)
- Advanced Diploma in Chemistry (AdvDip)
- Postgraduate Diploma in Chemistry (PGDip)
- Master of Applied Science in Chemistry (MAppSc)
- Doctor of Philosophy in Chemistry (PhD)

Qualification	Qualification Code	Important Dates			SAQA NLRD ID
		1st Offered	Phased Out Year*	Final Year Offered	
Diploma in Analytical Chemistry	DIACHI NDANFI	Jan 2017		2022	98010
ND: Analytical Chemistry	NDACH2		2016	2022	72210
Advanced Diploma in Chemistry	ADCHEI	Jan 2020			111354
PGDip in Chemistry	PDGCHI	Jan 2021			117947
MAppSc (Chemistry)	MSCMSI				96824
PhD Chemistry (Chemistry)	DPCMSI				96807

* The phased out period for 3 year programmes is 5 years from the phased out year (inclusive). The phased out period for 1 year programmes is 2 years from the phased out year (inclusive). No first time entry students will be allowed to register for phased out qualifications. DUT RULE G29 applies.

4. DIPLOMA IN ANALYTICAL CHEMISTRY (DIACHI)

Purpose of Qualification

Graduates of the diploma would be able to display competence in the application of knowledge in a broad range of varied work activities associated with a career in the Chemical and Allied Industries involving detergents, petroleum, plastics, food, pharmaceuticals, mining, water treatment, metallurgy. Graduates are required for specific practical applications such as quality control and testing or theoretical fields such as research and development. Quality control and assurance is an especially important field where industries need to verify their standards of operation and quality of manufactured materials.

The course is designed to meet the human resource needs of the chemical and allied industries and tertiary educational institutions, by providing nationally and internationally recognised tertiary education and training to students. Students who are successful in this programme may progress to the Advanced Diploma in Chemistry followed by a Postgraduate Diploma in Chemistry in order to gain access to a Masters or Doctoral qualification.

Apart from the formal course and practical work at DUT's Chemistry laboratories, the course also includes nine months Work Integrated Learning (WIL) component in an industrial setting, where skills and knowledge acquired at the university, are integrated and applied under real-life working conditions.

4.1 PROGRAMME STRUCTURE (3 YEAR)

Table 4.1 Programme Structure

Code	Subject Offering	Semester	Compulsory /Elective	Assessment Method	SAQA Credits	NQF Level	Prerequisite Subjects	Co-req
GINCI01	GENERAL AND INORGANIC CHEMISTRY	1a	C	Ex	16	5		
MTHCI01	MATHEMATICS I	1a	C	CA	16	5		
PHICI01	PHYSICS I	1a	C	Ex	16	5		
CSTNI01	Cornerstone 101 (IGE)	1a	C	CA	12	5		
CPUSI01	COMPUTER SKILLS I	1a	C	CA	8	5		
OPCHI01	ORGANIC AND PHYSICAL CHEMISTRY	1b	C	Ex	16	5		
ANCHI01	ANALYTICAL CHEMISTRY I	1b	C	Ex	20	6	GENERAL AND INORGANIC CHEMISTRY	
CCNSI01	COMMUNICATION SKILLS	1b	C	CA	8	5		
ITCHI01	Choice of 2 of the following electives (IGE): • Introduction to Technopreneurship • Cultural Diversity • Values in the Workplace • Sustainable Earth Studies	1b	E	CA	8	5		
CLDV101		1b	E	CA	8	5		
VWKP101								
SERS101								
ANCH201	ANALYTICAL CHEMISTRY II	2a	C	Ex	24	6	ANALYTICAL CHEMISTRY I	
APICI01	APPLIED INORGANIC CHEMISTRY	2a	C	Ex	12	6	GENERAL AND INORGANIC CHEMISTRY	
APOCI01	APPLIED ORGANIC CHEMISTRY	2a	C	Ex	12	6	ORGANIC AND PHYSICAL CHEMISTRY	
APPCI01	APPLIED PHYSICAL CHEMISTRY	2a	C	Ex	12	6	ORGANIC AND PHYSICAL CHEMISTRY	MATHEMATICS I
LDSHI01	Choice of 1 of the following electives (IGE): • Leadership • Constitutional Law & Human Rights	2a	E	CA	8	5		
CLHR101								
ACCH301	ANALYTICAL CHEMISTRY III: CHROMATOGRAPHY#	2b	C	Ex	12	6	ANALYTICAL CHEMISTRY II	
ACAS301	ANALYTICAL CHEMISTRY III: ATOMIC SPECTROSCOPY#	2b	C	Ex	12	6	ANALYTICAL CHEMISTRY II	
ACMS301	ANALYTICAL CHEMISTRY III: MOLECULAR SPECTROSCOPY#	2b	C	Ex	12	6	ANALYTICAL CHEMISTRY II	
ACEC301	ANALYTICAL CHEMISTRY III: ELECTROANALYTICAL CHEMISTRY#	2b	C	Ex	12	6	ANALYTICAL CHEMISTRY II	
CQLA101	CHEMICAL QUALITY ASSURANCE	2b	C	Ex	16	6	ANALYTICAL CHEMISTRY I	
EXLN101	EXPERIENTIAL LEARNING I (THEORY)#	3 a/b	C	CA	12	6	ANALYTICAL CHEMISTRY III: CHROMATOGRAPHY ATOMIC SPECTROSCOPY MOLECULAR SPECTROSCOPY ELECTROANALYTICAL CHEMISTRY	
CHPJ101	CHEMISTRY PROJECT I (THEORY)#	3 a/b	C	CA	36	6	ANALYTICAL CHEMISTRY III: CHROMATOGRAPHY ATOMIC SPECTROSCOPY MOLECULAR SPECTROSCOPY ELECTROANALYTICAL CHEMISTRY	
ASCE101	Community Engagement and Development (FGE)	3 a/b	C	CA	12	6		
EXLN201	EXPERIENTIAL LEARNING II (PRAC)#	3	C	CA	48	6		
CHPJ201	CHEMISTRY PROJECT II (PRAC)#	3	C	CA	12	6		

KEY:

* Assessment: Ex = examinable; CA = Continuous Assessment

**Numbers 1 to 3 indicates the year of study, "a"= Semester 1, "b"=Semester 2 (eg 2b=Second year, Semester 2)

These subjects are final level subjects.

A Pre-Req (prerequisite) means this subject must be passed prior to registration for the subsequent subject. A

Co-Req (corequisite) means these subjects must be registered and passed simultaneously.

FGE=Faculty General Education Module, IGE=Institutional General Education Module

4.2 PROGRAMME INFORMATION

4.2.1 Academic Integrity

Refer to the DUT General Rules pertaining to academic integrity GI3 (1) (o)

– covering falsification of academic records, plagiarism and cheating. These will be enforced wherever necessary to safeguard the worthiness of our qualifications, and the integrity of the Faculty of Applied Sciences at DUT.

4.2.2 Code of Conduct for Students

A professional code of conduct pertaining to behaviour, appearance, personal hygiene and dress shall apply to all students registered with the Faculty of Applied Sciences, at all times. Refer to Programme Rule 4.3.8 below.

4.2.3 Attendance

Students are expected to attend all planned academic activities as these are designed to provide optimal support for the required competency. Students are expected to be punctual for all academic activities. Penalties may be applied for late or poor attendance.

4.2.4 Work Integrated Learning (WIL)

The final year is focused on Work Integrated Learning (WIL) (Table 4.1 Programme Structure refers). Students will be required to register simultaneously in either January or July of each year for both the theory and practical modules, and prior to starting placement in industry.

Registration:

Students will register for the following semesterised theoretical modules in either January or July each year:

Theoretical Semester Modules:

- Experiential Learning I (EXLN101)
12 SAQA Credits
- Chemistry Project I (CHPJ101)
36 SAQA Credits
- Community Engagement and Development (ASCE101)
12 SAQA Credits

Students will only register for the following practical modules only when placed in industry:

Practical Modules:

- Experiential Learning II (WIL) (EXLN201)
48 SAQA Credits
- Chemistry Project II (WIL) (CHPJ201)
12 SAQA Credits

The theoretical lectures for Community Engagement and Development (ASCE101) and Experiential Learning I (EXLN101) will be conducted during the two weeks, following registration in January/July. Lectures for Chemistry Project I (CHPJ101) will thereafter be offered on one day per week for thirteen weeks each, during the first and second semester.

Students will complete the two practical modules over the remaining period for the year in industry. Registrations for students who have not completed practical modules will be rolled over to the following year.

- Experiential Learning I - where the student will acquire the needed soft skills for WIL.
- Chemistry Project I - where the student will be acquainted with research methods/techniques of undertaking literature surveys in preparation for Chemistry Project II in industry.
- Community Engagement and Development Module (FGE) - where the student will be given an appropriate assignment based on the subject material. (This module is independent of the WIL component.)
- Experiential Learning II - where the student will be placed in industry and trained in appropriate instrumental techniques.
- Chemistry Project II – where the student will undertake a project under the supervision of both a DUT mentor and an appropriately qualified Chemist/Analyst. The project will be based on a topic of relevance to the company and will be done in addition to the normal duties required by the company during the WIL period. The student will be required to present his/her findings by means of an oral presentation and a written report at DUT at the end of the period. (Approved by Senate Rules comm. 14/11/18)

WIL forms part of and is integral to the exit level outcomes of the Diploma in Analytical Chemistry. As highlighted in the CHE, WIL Good Practice Guidelines (2010) assessment will occur throughout the WIL period. Assessment and evaluation of the WIL component will be performed by university staff, workplace supervisors and examiners. Logbooks, assignment reports, projects, presentations or any other agreed evidence portfolios will be used to assess and evaluate student learning.

4.2.5 Assessment and Moderation

Students are expected to work steadily through the period during which they are registered in order to achieve the highest results possible.

- Assessment details are listed under each subject in this handbook.
- Assessments could include a variety of testing methods including, but not limited to, written tests, oral tests, theoretical and/or practical examinations, group work and assignments.
- Assignments must be submitted by the due date. Late submission will be penalised, unless a valid reason is provided.
- In the case of a continuous assessment subject (a subject which has no final examination/s or supplementary examination/s) opportunities for reassessment are provided for students who fail assessments. These are stipulated in the relevant study guide.
- Moderation follows the DUT Assessment Policy stipulations.
- Refer to Programme Rule 4.3.9 below.

4.2.6 Employment Opportunities

There is a demand for graduates in this field and industries are supportive of graduates pursuing further educational qualifications.

Graduates may also apply for associate membership of the South African Chemical Institute (SACI). SACI is associated with the South African Council for Natural Scientific Professions (SACNASP). The following link refers: <http://www.sacnasp.org.za/about-us/voluntary-associations.html>

4.2.7 Medical Conditions

Students must please note that the following medical conditions may preclude employment in some chemical industries: asthma, diabetes, allergies (skin), poor eyesight and colour blindness, epilepsy and hearing problems, or any medical conditions that may cause endangerment in the laboratories.

4.3 PROGRAMME RULES

4.3.1 Minimum Admission Requirements

In addition to DUT Rule G7, the following minimum entrance requirements and the selection criteria outlined in Rule 4.3.2 will apply for applicants with reference to:-

4.3.1.1 Academic Achievement

In line with the above, the applicants' school leaving academic achievement must comply with one of the following subjects at the stated minimum ratings as outlined in the table below:

- A National Senior Certificate (NSC) with endorsement for a diploma/degree:
- A Senior Certificate (SC) with the following subjects:
- A National Certificate Vocational (NCV) Level 4 with statutory

requirements for a diploma:

Compulsory Subjects	NSC	SC		NCV
	Rating	HG	SG	
Mathematics	4	D	B	60%
English	4	D	B	50%
Physical Science	3	E	C	50%

- Applicants that also obtained the Higher Certificate in Applied Sciences with Applied Science option (option 1). Applied Science (Chemistry and Statistics for the Applied Sciences) will be considered.

4.3.2 Selection Criteria

i) First Time Entering Applicants

In addition to the Minimum Admission Requirements (Rule 4.3.1), the following selection process will determine acceptance into the programme:

- All applicants must apply through the Central Applications Office (CAO).
- Initial shortlisting for selection is based on the applicant's academic performance in Grade 12 (Grade 11 or Grade 12 trial marks will be used for current matriculants).

Selection will be based on the ranking of applicants who meet the minimum requirements.

ii) Applicants entering via Articulation

Applicants holding a Higher Certificate in Applied Sciences who apply to articulate into the Diploma, will be selected for entry using a ranking system based on their aggregate marks. The number of students accepted into the programme will be capped at a maximum of 10 students.

4.3.3 Requirements

In addition to rule G12(1) a sub-minimum of 40% is required for the practical component of all subjects in which the semester mark is made up of theory and practical components. These are indicated in Table 4.1 Programme Structure.

4.3.4 Promotion to a Higher Level/Progression Rules

4.3.4.1 Promotion from Year 1 to Year 2:

The DUT Rule G16 applies.

4.3.4.2 Promotion from Year 2 to Year 3:

In accordance with the DUT Rule G16, students must pass all Analytical Chemistry III modules to proceed to Year 3 (WIL). Students are permitted to carry only one module from the preceding years into Year 3

4.3.5 Exclusion Rules

In addition to DUT Rule G17, a first semester student who fails three or more subjects with a final result of less than 40% in each of the failed subjects is not permitted to reregister in this programme. Deregistration from any subjects is subject to the provision of DUT Rule G6.

4.3.6 Interruption of Studies

In accordance with DUT Rule G21A(b), the minimum duration for this programme will be 3 years of registered study and the maximum duration will be 5 years of registered study, including any periods of WIL. Should a student interrupt their studies by more than three (3) years, the student will need to apply to the department for permission to reregister and will need to prove currency of appropriate knowledge prior to being given permission to continue with registration.

4.3.7 Work Integrated Learning Rules

The DUT Rule G28 applies.

4.3.8 Code of Conduct

In addition to the Student Code of Conduct in the DUT General Handbook for Students, and the relevant requirements as stated in the appropriate Study Guides, the following rules apply:

4.3.8.1 Dress Code and Conduct of Students in Practical Laboratory Facilities

Strict adherence to instructions issued by technical, laboratory or academic staff is required due to the need to ensure effective and safe practice in these facilities. Students must adhere to the dress code required during practical sessions. Misconduct or disregard for instructions will result in the student being denied access to the laboratory venue.

4.3.9 Attendance and Assessment

4.3.9.1 A student who, for any valid reason, is absent from planned academic activity must provide written proof of the reason for the absence to the lecturer concerned, within five (5) working days of returning to the institution in order to be considered for a special assessment.

4.3.9.2 The DUT Rule G13 (3) (a) which refers to special examinations also refers to special assessments set within departments for students who have missed coursework assessments. In these cases the department will determine the validity of the student's reason for not taking the assessment, and the nature of the special assessment.

4.3.10 Health and Safety

Students must adhere to all Health and Safety regulations both while at DUT and in WIL placements. Failure to do so will be treated as a breach of discipline. Refer to the appropriate Health and Safety policies.

4.3.11 General Education Modules

Students must comply with the university's General Education requirement. This includes the following standalone General Education modules which comprise of:

- 1 Compulsory DUT Cornerstone 101 module
- 1 Compulsory Faculty General Education module
- 3 Elective Institutional General Education modules (Students will select electives modules as indicated in Table 4.1 Programme Structure)

5. NATIONAL DIPLOMA: ANALYTICAL CHEMISTRY (NDACH2)

Purpose of Qualification

Graduates of the National Dip: Analytical Chemistry are able to apply the advanced scientific

principles and techniques of quantitative and qualitative analysis, quality control of raw materials and finished products, and research and development. Students benefit from the practical hands-on laboratory skills component with state-of-the-art equipment as well as direct exposure to the work situation.

This course is designed to meet the human resource needs of the chemical and allied industries and tertiary educational institutions, by providing nationally and internationally recognised tertiary education and training to students. This qualification allows entry to the Advanced Diploma in Chemistry.

5.1 PROGRAMME STRUCTURE (3 YEAR)

Table 5.1 Programme Structure

Code	Subjects	Assessment	Semester of Study	NATED Credits	Prerequisite Subjects	Corequisite Subjects
CHEM102	Chemistry I	Ex	1a	0.125		
MATH101	Mathematics I	CA	1a	0.083		
PHSA102	Physics I	Ex	1a	0.083		
CSK1103	Communication Skills I	CA	1a	0.083		
CSCC101	Computer Skills I	CA	1a	0.083		
ACHM103	Analytical Chemistry I	Ex	1b	0.125	Chemistry I	
INCH201	Inorganic Chemistry II	Ex	1b	0.100	Chemistry I	
ORCH201	Organic Chemistry II	Ex	1b	0.100	Chemistry I	
PHCH201	Physical Chemistry II	Ex	1b	0.100	Chemistry I	Mathematics I
ACHM204	Analytical Chemistry II	Ex	2a	0.100	Analytical Chemistry I	Analytical Chemistry Practical II
ACPR201	Analytical Chemistry: Practical II	Ex	2a	0.100	Analytical Chemistry I	
INCH301#	Inorganic Chemistry III	Ex	2a	0.139	Inorganic Chemistry II	
ORCH302#	Organic Chemistry III	Ex	2a	0.139	Organic Chemistry II	
PHCH301#	Physical Chemistry III	Ex	2a	0.139	Physical Chemistry II	
ACHM313#	Analytical Chemistry III (Module I)	Ex	2b	0.100	Analytical Chemistry II	
ACHM323#	Analytical Chemistry III (Module II)	Ex	2b	0.100	Analytical Chemistry II	
ACPR312#	Analytical Chemistry: Practical III (Mod I)	CA	2b	0.100	Analytical Chemistry II Analytical Chemistry Practical II	
ACPR322#	Analytical Chemistry: Practical III (Mod II)	CA	2b	0.100	Analytical Chemistry II Analytical Chemistry Practical II	
CQAS201	Chemical Quality Assurance	Ex	2b	0.100	Analytical Chemistry I	
EXAN101	Chemical Industry Practical (Experiential Learning)	CA	3	0.500	Analytical Chemistry III Analytical Chemistry: Practical III	
CMPJ301#	Chemistry Project III	CA	3	0.500	Analytical Chemistry III Analytical Chemistry: Practical III Experiential Learning	

KEY:

* Assessment: Ex = examinable; CA = Continuous Assessment

**Numbers 1 to 4 indicates the year of study, "a"= Semester 1, "b"=Semester 2 (eg 2b=Second year, Semester 2) # These subjects are final level subjects.

A Pre-Req (prerequisite) means this subject must be passed prior to registration for the subsequent subject. A Co-Req (corequisite) means these subjects must be registered and passed simultaneously.

5.2 PROGRAMME INFORMATION

5.2.1 Academic Integrity

Refer to the DUT General Rules pertaining to academic integrity G13(1)(o) – covering falsification of academic records, plagiarism and cheating. These will be enforced wherever necessary to safeguard the worthiness of our qualifications, and the integrity of the Faculty of Applied Sciences at DUT.

5.2.2 Code of Conduct for Students

A professional code of conduct pertaining to behaviour, appearance, personal hygiene and dress shall apply to all students registered with the Faculty of Applied Sciences, at all times. Refer to Programme Rule 5.3.8.

5.2.3 Attendance

Students are expected to achieve 100% attendance for all planned academic activities as these are designed to provide optimal support for the required competency. Students are expected to be punctual for all academic activities. Penalties may be invoked for late attendance. Refer to Programme Rule 5.3.9.

5.2.4 Work Integrated Learning (WIL)

Work Integrated Learning (WIL) comprises twelve months of experiential learning under the supervision of a qualified Chemist/Analyst. This is made up of two separate components, namely, Chemical Industry Practical (CIP) and Chemistry Project III (CP3). These components are treated as separate entities and are assessed separately.

During the first six months the student will be registered at DUT for CIP and carry out a variety of different techniques and procedures, which will be identified by the industrial supervisor in consultation with a mentor from DUT. At least one instrumental technique specified in the Experiential Learning (CIP) manual should be undertaken at the workplace. The student will be required to keep a record of his/her daily activities and at the end of the semester, compile a report (or portfolio) which is assessed by the industrial supervisor and DUT mentor.

During the second six months the student is required to register at DUT for Chemistry Project III, whilst working in industry. In addition to his/her normal duties, the student will be required to complete a project on some topic of relevance to the company and compile a written report. The project should include at least one instrumental technique that the student has used in CIP. At the end of the semester the student will be required to present his/her findings by means of an oral, poster presentation and a written report at DUT. Refer to Programme Rule 5.3.7.

5.2.5 Assessment and Moderation

Students are expected to work steadily through the period of registration in order to achieve the highest results possible.

- Assessment details are listed under each subject at the back of this handbook.
- Assessments could include a variety of testing methods including, but not limited to, written tests, oral tests, theoretical and/or practical examinations, group work and assignments.
- Assignments must be handed personally to the lecturer who will record their receipt. Late submission will be penalised.
- In the case of a continuous assessment subject (a subject which has no final examination/s or supplementary examination/s) opportunities for reassessment are provided for students who fail assessments. These are stipulated in the relevant study guide.
- Moderation follows the DUT Assessment Policy stipulations. Refer to Programme Rule 5.3.9.

5.2.6 Employment Opportunities

Employment may be found in a laboratory or production process as well as chemical and laboratory sales. Industries such as detergent, petroleum, plastics, food, pharmaceuticals, mining, water treatment, metallurgy and educational institutions employ graduates from this course. Graduates may work in a practical application such as quality control and testing, or a theoretical field such as research and development, with chemists and other technologists or technicians. Quality control and assurance is a field with a growing demand for these graduates. Opportunities exist for graduates to pursue further educational qualifications. Graduates may apply for associate membership of South African Chemical Institute (SACI).

5.2.7 Medical Conditions

Students must please note that the following medical conditions may preclude employment in some chemical industries: asthma, diabetes, allergies (skin), poor eyesight and colour blindness, epilepsy and hearing problems.

5.3 PROGRAMMERULES

5.3.1 Minimum Admission Requirements

In addition to DUT Rule G7, the following minimum entrance requirements and the selection criteria outlined in 5.3.2 apply for applicants with reference to:

- **Academic Achievement**

In line with the above, applicants' school leaving academic achievement must comply with one of the following:-

- i) a National Senior Certificate (NSC) with endorsement for diploma/degree entry with the following subjects at the stated minimum ratings

Compulsory Subject	NSC Rating
English	4
Mathematics	4
Physical Science	4
One 20 credit subject	3

- ii) a Senior Certificate is matriculation exemption with the following subjects at the stated minimum ratings

Compulsory Subjects	HG	SG
Mathematics	D	B
Physical Science	D	B

- iii) a National Certificate (Vocational) Level 4 with statutory requirements for a diploma entrance and the following subjects at the stated minimum ratings
(Approved by Senate wef 2013/08)

Compulsory Subjects	Minimum
English	50%
Mathematics	50%
Physical Sciences (or recognised equivalent)	60%

- **Admission Requirements based on Work Experience, Age and Maturity; and Recognition of Prior Learning**

The DUT Rules G7 (3), and G7 (8) respectively, will apply. (Approved by Senate Rules Comm wef 2014/10)

- **Admission of International Students**

a. The DUT's Admissions Policy for International Students and DUT Rules G4 and G7 (5) will apply.

b. International students must meet the equivalent programme minimum entrance requirements as stated above.

(Approved by Senate Rules Comm wef 2014/10)

- **Admission of Students from other Institutions**

In addition to the relevant DUT Rules a transferring student will only be accepted if there are places available and the student has met the applicable entrance requirements of the university. (Approved by Senate Rules Comm wef 2014/10)

5.3.2 Selection Criteria

In addition to the Minimum Admission Requirements (Rule 5.3.1), the following selection process will determine placement in the programme:

- All applicants must apply through the Central Applications Office (CAO).
- Initial shortlisting for selection is based on the applicant's academic performance in Grade 12 (Grade 11 or Grade 12 trial marks will be used for current matriculants).
- Applicants who meet the above criteria will be ranked based on performance in grade 12 final marks.

Provisional acceptance is given to selected applicants awaiting National Senior Certificate* (NSC) results. If the final Grade 12 NSC* results do not meet the minimum entrance requirements, this provisional acceptance will be withdrawn.

- Final selection for placement will be based on results of the ranking process, as well as available places (refer to DUT Rule G5).
(or SC / NC (V)) (Approved by Senate Rules Comm wef 2014/10)

5.3.3 Pass Requirements

- In addition to rule G12(1) a sub-minimum of 40% is required for the practical component of all subjects in which the semester mark is made up of theory and practical components. These are indicated in Table 5.1 Programme Structure.
(Approved by Senate Rules Comm wef 2014/10)

5.3.4 Promotion to a Higher Level/Progression Rules

5.3.4.1 Promotion from Year 1 to Year 2

The DUT Rule G16 applies.

5.3.4.2 Promotion from Year 2 to Year 3

In addition to DUT Rule G16, students will be promoted to Year 3 provided that they have no more than one module outstanding from previous years.

5.3.5 Exclusion Rules

In addition to DUT Rule G17, a first semester student who fails three or more subjects with a final result of less than 40% in each subject is not permitted to reregister in this programme. Deregistration from any subjects is subject to the provision of DUT Rule G6. *(Approved by Senate Rules Comm wef 2014/10)*

5.3.6 Interruption of Studies

In accordance with DUT Rule G21A(b), the minimum duration for this programme will be 3 years of registered study and the maximum duration will be 5 years of registered study, including any periods of WIL. Should a student interrupt their studies by more than three (3) years, the student will need to apply to the department for permission to reregister and will need to prove currency of appropriate knowledge prior to being given permission to continue with registration. *(Approved by Senate Rules Comm wef 2014/10)*

5.3.7 Work Integrated Learning Rules

The DUT Rule G28 applies. *(Approved by Senate Rules Comm wef 2014/10)*

5.3.8 Code of Conduct

In addition to the Student Code of Conduct in the DUT General Handbook for Students, and the relevant requirements as stated in the appropriate Study Guides, the following rules apply:

5.3.8.1 **Conduct of Students in Practical Facilities**

Strict adherence to instructions issued by technical, supervisory or academic staff is required due to the need to ensure effective and safe practice in these facilities. Misconduct or disregard for instructions will be referred to the relevant disciplinary procedure. *(Approved by Senate Rules Comm wef 2014/10)*

5.3.8.2 **Uniforms**

Students must adhere to instructions issued by technical, supervisory or academic staff regarding the specific dress code required during practicals. Non-compliance will result in the student being denied access to the venue. *(Approved by Senate Rules Comm wef 2014/10)*

5.3.9 Attendance and Assessment

A student who, for any valid reason is absent from a particular practical or test, must provide written proof of the reason for the absence to the lecturer concerned, within **five (5) working** days of returning to the institution in order to be considered for a special assessment. *(Approved by Senate Rules Comm wef 2014/10)*

The DUT Rule G13 (3) (a) which refers to special examinations also refers to special assessments set within departments for students who have missed coursework assessments. In these cases the department will determine the validity of the student's reason for not taking the assessment, and the nature of the special assessment.

(Approved by Senate Rules Comm wef 2014)

5.3.10 Health and Safety

Students must adhere to all Health and Safety regulations both while at DUT and in WIL placements. Failure to do so will be treated as a breach of discipline. Refer to the appropriate Health and Safety policies.

(Approved by Senate Rules Comm wef 2014/10)

6. ADVANCED DIPLOMA IN CHEMISTRY (ADCHEI)

Purpose of Qualification

The Advanced Diploma in Chemistry (**ADCHEI**) represents a level of qualification that recognizes the ability to gain and apply a range of new knowledge, skills and understanding designed not only to meet the needs of various chemical industries, but also to provide the most appropriate preparation for further study. The curriculum therefore builds upon and advances the fundamental areas of study in chemistry: organic, physical, inorganic and analytical chemistry, along with a rich set of research- focused and instrumentation intensive laboratory experiences aligned with those areas. Graduates can obtain employment as chemical laboratory technologists/chemists in quality control, research and development or teaching laboratories. This course allows entry into the Postgraduate Diploma in Chemistry.

6.1 PROGRAMME STRUCTURE

Table 6.1 Programme Structure

Module Name	Module Code	Study Level	NQF Level	Module Credits	C/E	Pre-Req	Co-Req	Stream	Exam
Inorganic Chemistry IV	INOC401	Ia	7	20	C	Nil	Nil	N/A	Yes
Organic Chemistry IV	ORGC401	Ia	7	20	C	Nil	Nil	N/A	Yes
Analytical Chemistry IV	ANAC401	Ia	7	32	C	Nil	Nil	N/A	Yes
Physical Chemistry IV	PHYC401	Ib	7	20	C	Nil	Nil	N/A	Yes
Chemistry Project IV	CHEP401	Ib	7	32	C	Nil	Nil	N/A	No (CA)
Total credits for Graduation (minimum): 124 Key: Ia = first semester; Ib = second semester; C = compulsory; E = Elective;									

6.2 DURATION OF PROGRAMME

In accordance with DUT Rules G21C(2) and G21C(3), the duration of the programme is as follows:

The minimum duration is one year of full time study.

6.3 PROGRAMME RULES

6.3.1 Minimum Admission Requirements

In addition to DUT rules G7 and G21C(1), applicants must be in possession of the following minimum admission entry requirements into this programme:

- National Diploma in Analytical Chemistry; or
- Diploma in Analytical Chemistry; or
- A recognized equivalent qualification at NQF Level 6.

6.3.2 Selection Criteria

Entry into the Advanced Diploma in Chemistry programme is not automatic. A limited number of places are available to students and selection will be on the basis of academic performance as determined by a ranking system, based on their performance in all S3 and S4 modules.

6.3.3 Pass Requirements

In addition to the DUT Rules G12, G14 and G15, unless otherwise specified, where practicals are constituted as part of the course mark, a sub-minimum of 40% is applicable to the practical mark.

6.3.4 Promotion to a Higher Level/Progression Rules

The DUT Rule G16 applies.

Exclusion Rules

In DUT Rules G17 and G21C(3) apply.

6.3.5 Interruption of Studies

The DUT Rule G6B pertaining to interruption of studies will apply.

7. POSTGRADUATE DIPLOMA IN CHEMISTRY (PGDCHI)

Purpose of Qualification

The Postgraduate Diploma in Chemistry (PGDCHI) serves the purpose of allowing for specialisation in Chemistry. It extends and deepens knowledge and skills attained in the Advanced Diploma, whilst focusing on enabling working professionals to apply critical thinking in more complex and contemporary fields of study. In addition, the Postgraduate Diploma in Chemistry prepares students for further research-based postgraduate study with an all-inclusive supervised research component leading to research outputs. This qualification allows entry into the M AppSc degree in Chemistry.

7.1 PROGRAMME STRUCTURE

Table 7.1 Programme Structure

Code	Module Full-Time Option	Study Level#	NQF Level	HEQSF Credits	Compulsory/Elective	Pre-Requisite	Assessment
INCH801	Inorganic Chemistry V	Ia	8	20	C	N/A	EX
ORCH801	Organic Chemistry V	Ia	8	20	C	N/A	EX
ANCH801	Analytical Chemistry V	Ia	8	20	C	N/A	EX
SCGY801	Scientific Methodology	Ia	8	8	C	N/A	CA*
PHCH802	Physical Chemistry V	Ib	8	20	C	N/A	EX
CHPR802	Chemistry Project V	Ib	8	32	C	Scientific Methodology	CA*

Total number of credits for Graduation 120

*CA = Modules are Continuously Assessed

#Study level: Ia = Year I, Semester I and Ib = Year I, Semester 2

A Pre-Req (prerequisite) means this module must be passed prior to registration for the subsequent subject.

7.2 PROGRAMME INFORMATION AND DURATION

Students will register full-time and complete the modules in one year as in Table 9.1. The DUT Rules G22B(2) and G22B(3) apply.

7.3 PROGRAMME RULES

7.3.1 Minimum Admission Requirements

In addition to DUT rules G7 and G22B, applicants must be in possession of the following minimum admission entry requirements into this programme:

7.3.1.1 Advanced Diploma in Chemistry;

7.3.1.2 or a recognized equivalent qualification at NQF Level 7.

7.3.2 Selection Criteria

In addition to the Minimum Admission Requirements, final selection will be based on the following:

- i) Applicants who meet the minimum requirements will be ranked based on academic performance.
- ii) The number of available places in the programme. (Refer to DUT Rule G5).
- iii) Pass Requirements

In addition to the DUT Rules G12, G14 and G15, unless otherwise specified, where practicals are constituted as part of the course mark, a sub-minimum of 40% is applicable to the practical mark.

7.3.3 Promotion to a Higher Level/Progression Rules

The DUT Rule G16 applies.

7.3.4 Exclusion Rules

The DUT Rules G17 and G22B (3) apply.

7.3.5 Interruption of Studies

The DUT Rule G6 (2) applies.

8. MASTER OF APPLIED SCIENCE IN CHEMISTRY (MSCMSI)

Purpose of Qualification

The purpose of the Masters qualification is aimed at the different areas of chemistry research coupled with high-end instrumental analysis. The Masters candidate is required to demonstrate the ability to make an informed decision on the appropriate methodology, choice of analytical instruments and to communicate ideas and results scientifically.

8.1 PROGRAMME STRUCTURE

Subject	Subject Description	Period of Study	Assessment Method	NATED Credits
RTCM511	Research Thesis (1 st Reg)	Annual	Research Project	1.000
RTCM521	Research Thesis (2 nd Reg)			
RTCM531	Research Thesis (3 rd Reg)			
RTCM541	Research Thesis (4 th Reg)			

8.2 PROGRAMME INFORMATION

The curriculum consists of a research project and dissertation.

8.3 PROGRAMME RULES

Postgraduate Publications Requirements for Completion of the Degree

Master's students are required to have:-

ONE publication submitted to a DHET approved journal prior to completion of the degree.

8.4 MINIMUM ADMISSION REQUIREMENTS

In addition to Rule G24 (1) applicants must be in possession of a Post Graduate Diploma in Chemistry or BTech Chemistry degree or equivalent qualification.

8.5 DURATION

As per Rule G24 (2).

9. DOCTOR OF PHILOSOPHY IN CHEMISTRY (DPCMSI)

Purpose of Qualification

The purpose of the doctoral qualification is to develop new knowledge and research that is novel, innovative and technologically driven. The doctoral candidate is required to demonstrate high level research capability and to make a significant and original academic contribution to solving societal problems.

9.1 PROGRAMME STRUCTURE

Subject	Subject Description	Period of Study	Assessment Method	NATED Credits
RTCM61I	Research Thesis (1 st Reg)	Annual	Research Project	1.000
RTCM62I	Research Thesis (2 nd Reg)			
RTCM63I	Research Thesis (3 rd Reg)			
RTCM64I	Research Thesis (4 th Reg)			
RTCM65I	Research Thesis (5 th Reg)			

9.2 PROGRAMME INFORMATION

The curriculum consists of a research project and dissertation.

9.3 PROGRAMME RULES

Postgraduate Publications Requirements for Completion of the Degree

Doctoral students are required to have: -

- 9.3.1 ONE publication accepted by a DHET approved journal; and
- 9.3.2 ONE publication submitted to a DHET approved journal, prior to completion of the degree

9.4 MINIMUM ADMISSION REQUIREMENTS

In addition to Rule G25 (1) applicants must be in possession of a MAppSc (Chemistry) degree or equivalent qualification.

9.5 DURATION

As per Rule G25 (2).

10. SERVICED SUBJECTS

The servicing department's rules apply to all serviced subjects. The following subjects are serviced externally to this department.

Servicing Department	Serviced Subject	Subject Code
Department of Mathematics	Mathematics I	MATH101
Department of Physics	Physics I	PHSA102
Department of English and Communication	Communication Skills I	CSK1103
Department of Information Technology	Computer Skills I	CSCC101
Centre for General Education	Cornerstone 101	CSTN101
Dept of Media, Language and Communication	Cultural Diversity	CLDV101
Dept of Entrepreneurial Studies	Introduction to Technopreneurship	ITCH101
Dept of Horticulture	Sustainable Earth Studies	SERS101
Dept of Human Resources Management	Values In The Workplace	VWKP101
Dept of Applied Law	Constitutional Law & Human Rights	CLHR101
Centre for General Education	Leadership	LDSH101
Dept of Food and Nutrition	Community Engagement & Development	ASCE101

The following subjects are serviced from this department:

SERVICED PROGRAMME	SUBJECT NAME	SUBJECT CODE
Bachelor of Health Science in Biomedical Technology	Chemistry	CMTR101
Bachelor of Health Science: Medical Care	Chemistry	CMTR101
Bachelor of Applied Science in Biotechnology	Chemistry I	CSRY101
ND: Chemical Engineering	Chemistry IA	CHEM102
	Organic Chemistry II	ORCH201
	Inorganic Chemistry II	INCH201
	Physical Chemistry II	PHCH201
Bachelor of Engineering Technology in Chemical Engineering	Engineering Chemistry IA	ENCA101
	Engineering Chemistry IB	ENCB101
	Engineering Chemistry IIA	ENCM201
Bachelor of Health Science in Clinical Technology	Chemistry	CMTR101
ND: Dental Technology	Physics and Chemistry I	PHCD121
ND: Emergency Medical Care and Rescue	Basic Science I (Annual)	BSCN101
ND: Environmental Health	Physics & Chemistry I (Annual)	PHCM111

Bachelor of Health Science in Environmental Health	Chemistry I	CHMT101
ND: Food and Consumer Science	Physical Science I (Annual)	PSCN101
Diploma in Consumer Sciences in Food and Nutrition	Chemistry I	APSC 101
Diploma in Consumer Sciences in Food and Nutrition	Chemistry I	APSC 101
Bachelor of Applied Science in Food Technology	Chemistry I	CSRY101
Bachelor of Health Science in Homeopathy	Chemistry I	CHHC101
Bachelor of Health Science in Chiropractic	Chemistry I	CHHC103
Higher Certificate in Applied Sciences	Chemistry I	CHEM102
ND: Pulp & Paper Technology	Chemistry IA	CHEM102
	Physical Chemistry II	PHCH201
Diploma:Pulp & Paper Technology	Chemistry A	CHME101
	Chemistry B	CHME102
	Chemistry C	CHME103
Bachelor of Health Sciences: Radiotherapy Nuclear Medicine Diagnostic Sonography Diagnostic Radiography	Chemistry	CSTY101
ND: Somatology	Science I : Chemistry	SCCH101
	Science II	SCCS201
Diploma in Textile Technology	Chemistry I	CHSY101
Bachelor of Applied Sciences: Textile Science	Chemistry	CHEM101
Bachelor of Health Sciences in Dental Technology	Chemistry	CHMS101

II. SHORT COURSES

The following short courses are currently offered by the Department.

- General Laboratory Practice (3 days)
- Basic Gas Chromatography (3 days)
- Advanced Gas Chromatography (5 days)
- Basic Atomic Absorption Spectroscopy (3 days)
- Advanced Atomic Absorption Spectroscopy (5 days)

For further details, please contact:

Centre for Continuing Professional Education

(CCPE) Philiswa Charity Dlamini

Phone : 031 3736016 / 7

Email : philiswad@dut.ac.za

Or

Selisha Ramduth

Email : selishar@dut.ac.za

12. SUBJECT CONTENT

The following must be read in conjunction with the appropriate study guides.

12.1 DIPLOMA: ANALYTICAL CHEMISTRY (DIACHI) CPUS101 - COMPUTER SKILLS I

CONTACT TIME (Periods per week)

Theory and demonstration of practical (3)

ASSESSMENT

Continuous Assessment

Theory Test:	25%
Practical evaluation presentations (or a test):	25%
Word processing:	25%
Spreadsheets:	25%

The practical tests will be set in the students' practical time. (Each group will have their own test).

A make-up theory test will be set at the end of the semester. This would be for students who missed the theory test.

SYLLABUS: Introductory theory on Information and communications technology,

MS Word, MS Excel, MS PowerPoint, Introduction to Internet search engine/s, Introduction to e-Mail

CSTN101 – CORNERSTONE 101

CONTACT TIME (Periods per week)

Theory (3); small group activity & independent study

ASSESSMENT

Continuous Assessment

A weekly blog written by each student	20%
Tutorial attendance (forfeited if student attends Less than 80% of tutorials)	10%
Visual artefact	15%
Written report	30%
Oral presentation	5%
Peer assessment.	10%

SYLLABUS: The module content will be developed around the concept of journeys, across time, across space, and across human relationships; the first use of the concept will take the journey of the uMgeni River (which is close to all DUT campuses) as a metaphor. The module will start with the analysis of a particular issue or metaphor (one critical event or development will be analysed; the event in focus will be selected on the basis of its connections to the theme of journeys and its relevance to the issues of ethics, diversity and critical citizenry). The final section of the module will identify and integrate learning from earlier sections, and examine implications for further learning. At each stage of the module, students will be required to engage in activities that involve reflection and build communicative practices.

GINCI01- GENERAL AND INORGANIC CHEMISTRY

CONTACT TIME (Periods per week)

Theory (4); Practical (1 X 3 hr)

ASSESSMENT

Course Mark

Theory: 2 x 1 hour tests:	60%
Practicals: 2 x 3 hour tests (15% each)	30%
Computer Tutorials and Practical reports:	10%

Final Mark

Examination: 1 x 3 hour

paper

Final mark: Course mark (40%) + Examination mark

(60%)

SYLLABUS: Atomic structure and periodic table, types of bonding, reactions and stoichiometry, types of reactions, properties of s and p block elements.

MTHCI01-MATHEMATICS I

CONTACT TIME (Periods per week)

Theory (4); Tutorials (2)

ASSESSMENT

Continuous

Assessment

Best 3 out of 4 short (30 minute) tests and/or assignments:

20%

2 x 1.5 hour Major tests (moderated). One for each section:

80% Pass mark = 50%, with a sub-minimum of 40% for each major test. **Note:**

1. Students who obtain a final result between 45% and 49% will be eligible to write a 3-hour make-up test covering the whole syllabus, at the end of the semester, with the students who missed a control (major) test with a valid reason which, if passed, will result in a mark of 50% being allocated.
2. Students who get at least 50% but who fail due to the test sub-minimum also qualify for this test.
3. Students who miss a major test & are allowed to write the make-up test on the work covered in the missed test, **DO NOT** qualify for the above.

SYLLABUS: Algebraic functions, determinants of 2x2 and 3x3 'square' matrices, algebraic fraction into partial fractions, trig operations, algebraic functions and inverse functions, simple statistical examples, hyperbolic functions, simple algebraic, trig and hyperbolic functions

PHIC101 - PHYSICS I

CONTACT TIME (Periods per week)

Theory (4); Tutorial (2)

ASSESSMENT

Course Mark

Theory: Best 2 out of 3 theory tests (1 hour each):	65%
Practicals (15%) & Test (85%):	30%
Tutorial Tests:	5%

Final Mark

Examination: 1 X 3 Hour theory examination

Final mark: Course Mark (40%) + Examination Mark (60%)

SYLLABUS: Units and Vectors, Kinematics in One Dimension, Forces and Newton's Laws of Motion, Work and Energy, Impulse and Momentum, Elasticity, Fluids, Temperature & Heat, Electrostatics, Electric Circuits, The Reflection of Light: Mirrors, The Refraction of Light: Lenses and other media.

ANCH101 - ANALYTICAL CHEMISTRY I

CONTACT TIME (Periods per week)

Theory (4); Practical (2 x 3 hr)

ASSESSMENT

Course Mark

Theory: 2 x 1 hour tests:	60%
Practicals: 2 x 3 hour tests (15% each)	30%
Computer Tutorials and Practical reports:	10%

Final Mark

Examination: 1 x 3 hour paper

Final mark: Course mark (40%) + Examination mark (60%)

SYLLABUS: Introduction to Analytical Chemistry, elementary statistics, laboratory Practice and Safety, titrimetric Analysis, sampling and sample handling, gravimetric analysis, introduction to instrumental analysis, basic report writing.

CCNS101 - COMMUNICATION SKILLS I

CONTACT TIME (Periods per week)

Theory (2); Tutorial (1)

ASSESSMENT

Continuous Assessment

1 x 1 hour Theory test:	33.3%
1 x Oral Presentation:	33.3%
1 x Written Project:	33.3%

SYLLABUS: Communication theory; oral presentation; technical writing skills; group communication skills.

CLDVI01 – CULTURAL DIVERSITY

CONTACT TIME (Periods per week)

Theory (2); small group activity & independent study

ASSESSMENT

Continuous Assessment

Assignments:	20%
Oral presentation:	40%
Portfolio:	40%

SYLLABUS: The module will be introduced by defining culture and establishing the salience of culture in the local and global context. There is also some attention paid to diverse cultural groups in the SA and global context. The core content focuses on aspects of social responsibility and gives strong attention to issues of anti-discriminatory and anti-oppressive practices. Social justice is unpacked and the effect of marginalization on oppressed groups discussed. Consciousness raising and social action and dialoguing across differences is used to interweave the introductory and main aspects of the module. It forms an appropriate way to conclude the module as it requires students to engage in activities that involve reflection and personal commitment to anti-oppressive practices.

ITCHI01 – INTRODUCTION TO TECHNOPRENEURSHIP

CONTACT TIME (Periods per week)

Theory (2); small group activity and independent study

ASSESSMENT

Continuous Assessment

Tests:	50%
Individual Participation/Graduate Attributes:	10%
Group Written Assignment:	20%
Group Oral Presentation to Panel:	20%

SYLLABUS: Small Business and Entrepreneurship – the language, differences, need and statistics. Supporting organisations and policies; The entrepreneurial profile; Creativity, innovation, self-awareness and Technopreneurship; Introduction to business structures; Introduction to business functions (Marketing, Finance, HR & Operations); Introduction to the Feasibility Study and Business Plan.

OPCH101- ORGANIC AND PHYSICAL CHEMISTRY

CONTACT TIME (Periods per week)

Theory (4); Practical (1 x 3 hr)

ASSESSMENT Course Mark

Theory: 2 x 1 hour tests: 60%

Practical: Test and Practical report: 40%

Final Mark

Examination: 1 x 3 hour paper

Final mark: Course mark (40%) + Examination mark (60%)

SYLLABUS: Rates of reactions, equilibrium, acids, bases and neutralisation reactions. Colligative properties and electrochemistry. Organic chemistry involving saturated and unsaturated hydrocarbons, alcohols and phenols, ethers, aldehydes and ketones, carboxylic acids and esters, amines and amide

SERS101 – SUSTAINABLE EARTH STUDIES

CONTACT TIME (Periods per week)

Theory (2); small group activity & independent study

ASSESSMENT

Continuous Assessment

Tests: 30%

Essay: 30%

Report: 40%

SYLLABUS:

Introduction and significance

- Understanding the relevance, meaning and purpose of earth studies to personal and professional life.

The Biosphere in space

- Core concepts of the earth as a nurturing complete system
- Locating the earth as a living planet in the solar system, galaxy and space
- Describing the earth using mapping and cartography systems
- Establishing position on the earth's surface (lines of meridians and latitude, co-ordinates and GPS systems)

Earth systems (Abiotic or non-living component)

- Overview of the nature and significance of earth's structure (lithosphere), atmosphere and water systems (hydrosphere) using relevant global and local examples

- Bio-element recycling of major minerals and elements

Ecology (Biotic or living component)

- Core concepts of ecology, primary production and transfer of energy
- Connected webs of human, animal and plant communities
- Global and local example

Biodiversity

- Nature, importance and characteristics of biodiversity (the variety and distribution of life)
- Ecosystem and species biodiversity
- Biodiversity and its link to human wellbeing-
- Biodiversity of Durban - local wildlife and plants on our doorstep
- Biodiversity under threat - Climate Change, Species extinctions,
- Habitat degradation
- Overview of Global and local Responses to Biodiversity (Rio +20, COP 17, Local Agenda 21, DMOSS)
- Pressing issues in biodiversity conservation in KZN (Rhino poaching (Ezimvelo), cycad trading, alien plant invasions, unsustainable harvesting of muti plants)

VWKPI01 – VALUES IN THE WORKPLACE

CONTACT TIME (Periods per week)

Theory (2); small group activity & independent study

ASSESSMENT

Continuous Assessment

Two assignments:	40%
One oral presentation:	20%
Reflection:	20%
Peer Assessment:	20%

SYLLABUS:

The module will begin with a reflection on personal values and move to a discussion on how they intersect with values in the workplace. Small group discussions will be formed around how to build positive values in the workplace and the vital themes of ethics, respect, interconnectedness, honesty, creativity and human diversity will form the basis for building “sacred spaces at work.” This will set the tone to unpack issues around leadership values and ethics and ethical decision making. The final section of the module will integrate all these aspects and students will be required to identify the implications of what they have learnt to develop social responsibility and their roles as citizens.

ANCH201 - ANALYTICAL CHEMISTRY II

CONTACT TIME (Periods per week)

Theory (4); Practical (2 x 3 hr)

ASSESSMENT

Course Mark

Theory: 2 x 1 hour tests:	60%
Practical: Test and Practical report:	40%

Final Mark

Examination: 1 x 3 hour paper

Final mark: Course mark (40%) + Examination mark (60%)

SYLLABUS: Advanced titrimetric analysis, sampling and sample handling, gravimetric analysis, introduction to instrumental analysis

APIC101- APPLIED INORGANIC CHEMISTRY

CONTACT TIME (Periods per week)

Theory (3); Practical (4 sessions per semester)

ASSESSMENT

Course Mark

Theory: 2 x 1 hour tests: 60%

Practical: Test and Practical report: 40%

Final Mark

Examination: 1 x 3 hour paper

Final mark: Course mark (40%) + Examination mark (60%)

SYLLABUS: Theories of bonding; coordination chemistry; crystal field theory; descriptive chemistry of first transition series, associated elements and nuclear chemistry.

APOC101- APPLIED ORGANIC CHEMISTRY

CONTACT TIME (Periods per week)

Theory (3); Practical (4 sessions per semester)

ASSESSMENT

Course Mark

Theory: 2 x 1 hour tests: 60%

Practical: Test and Practical report: 40%

Final Mark

Examination: 1 x 3 hour paper

Final mark: Course mark (40%) + Examination mark (60%)

SYLLABUS: Acids and Bases in Organic Chemistry; Conformation of Molecules; Stereochemistry; Elimination and Substitution Reactions; Polymers; Aromatic Compounds; Aldehydes and Ketones; Dicarbonyl Compounds; Carbohydrates; Lipids; Amino Acids and Proteins; Spectroscopy.

APPC101- APPLIED PHYSICAL CHEMISTRY

CONTACT TIME (Periods per week)

Theory (3); Practical (4 sessions per semester)

ASSESSMENT

Course Mark

Theory: 2 x 1 hour tests: 60%

Practical: Test and Practical report: 40%

Final Mark

Examination: 1 x 3 hour paper

Final mark: Course mark (40%) + Examination mark (60%)

SYLLABUS: Thermodynamics, Chemical Kinetics, Change of Phase, Surface Chemistry, Electrochemistry

CLHR101 – CONSTITUTIONAL LAW AND HUMAN RIGHTS

CONTACT TIME (Periods per week)

Theory (2); small group activity & independent study

ASSESSMENT

Continuous

Assessment

Tests: 50%

A written group assignment on Constitutional Law 20%

A written individual assignment on Human Rights 20%

One oral presentation 10%

SYLLABUS:

Introduction; Basic Features of the Constitution; Constitutional History of South Africa; Constitutional principles; The Constitution as the Supreme Law and Source of Law; Organs of State; Traditional leaders; The role of the Constitutional Court; State Institutions supporting Constitutional Democracy.; Human Rights.; Introduction to The Bill of Rights; Bearers and Holders of Rights; Limitation of Rights; Application of the Substantive Provisions of the Bill of Rights to Human Rights issues in South Africa; Suitable relief for enforcement of Human Rights / Remedies for violation of Human Rights; Substantive provisions of the Bill of Rights; Case Studies & Practical Examples on Human Rights Issues.

LDSH101 – LEADERSHIP

CONTACT TIME (Periods per week)

Theory (2); small group activity & independent study

ASSESSMENT

Continuous Assessment

Participation in classes and in activities

Related to the community project: 25%

A written report (5 000 words) that reflects on the Experience of leadership in the community project 35%

Oral presentation in class seminars 10%

A weekly blog written by each student 20%

Mentor's report 10%

SYLLABUS:

Negotiation of ground rules; Developing of questions and goals for learning; Existing experience of leadership; Basic concepts and theories of leadership; Preparing for community engagement; Case studies of leadership – engagement with leaders from different contexts; Involvement in a specific community or workplace project; Applying theory in analysing the experience of leadership; Reflection on self and setting of goals; Evaluation

ACAS30I - ANALYTICAL CHEM III: ATOMIC SPECTROSCOPY

CONTACT TIME (Periods per week)

Theory (3); Practical (4 sessions per semester)

ASSESSMENT

Course Mark

Theory: 2 x 1 hour tests: 60%

Practical: Test and Practical report: 40%

Final Mark

Examination: 1 x 2 hour paper

Final mark: Course mark (40%) + Examination mark (60%)

SYLLABUS: Flame emission and atomic absorption spectrometry; Emission spectrometry and X-ray methods

ACCH30I - ANALYTICAL CHEM III: CHROMATOGRAPHY

CONTACT TIME (Periods per week)

Theory (3); Practical (8 sessions per semester)

ASSESSMENT

Course Mark

Theory: 2 x 1 hour tests: 60%

Practical: Test and Practical report: 40%

Final Mark

Examination: 1 x 2 hour paper

Final mark: Course mark (40%) + Examination mark (60%)

SYLLABUS: Introduction to analytical separations, Gas chromatography, High performance liquid chromatography.

ACEC30I – ANALYTICAL CHEM III: ELECTROANALYTICAL

CONTACT TIME (Periods per week)

Theory (3); Practical (4 sessions per semester)

ASSESSMENT

Course Mark

Theory: 2 x 1 hour tests: 60%

Practical: Test and Practical report: 40%

Final Mark

Examination: 1 x 2 hour paper

Final mark: Course mark (40%) + Examination mark (60%)

SYLLABUS: Potentiometry, Polarography, Electrogravimetry and Coulometry, Thermal methods of analysis.

ACMS30I ANALYTICAL CHEM III: MOLECULAR SPECTROSCOPY

CONTACT TIME (Periods per week)

Theory (3); Practical (4 sessions per semester)

ASSESSMENT

Course Mark

Theory: 2 x 1 hour tests: 60%

Practical: Test and Practical report: 40%

Final Mark

Examination: 1 x 2 hour paper

Final mark: Course mark (40%) + Examination mark (60%)

SYLLABUS: Introduction to absorption and emission spectrometry, Molecular spectroscopy (UV, Visible and IR).

CQLA10I- CHEMICAL QUALITY ASSURANCE

CONTACT TIME (Periods per week)

Theory (4); Tutorial (2)

ASSESSMENT

Course Mark

Theory: Tests: 2 x 1 hour tests (weighted 40% each): 80%

Assignment: 20%

Final Mark

Examination: 1 x 3 hour theory paper

Final Mark: Course Mark (40%) + Examination mark (60%)

SYLLABUS: Quality dimensions, selected codes of practice, basic statistics and SPC, good laboratory practice.

EXLN10I- EXPERIENTIAL LEARNING I (WIL)

ASSESSMENT

Continuous Assessment

Test 1 15%

Test 2 25%

Assignment on a soft skills topic 30%

EXLN Class Performance Rubic 30%

The Rubic mark is calculated in % after summing up the ratings as per spreadsheet – (Refer to the Chemistry WIL manual for the spreadsheet).

SYLLABUS: Report writing skills, CVs, Basic Statistics using Spread Sheets, Power Point Slides & Interpersonal skills.

CHPJ10I- CHEMISTRY PROJECT I (WIL)

ASSESSMENT

Continuous Assessment

Assignment 1 33%

Assignment 2 34%

Assignment 3

33%

If the student opts to improve his/her mark by doing an additional assignment, an average of the mark for the Assignment 2 and the “make up” Assignment will be counted as the mark for Assignment 2. Assignment difficulty: 3>1> Make-up > 2.)

SYLLABUS: Research Methodology and topics related to industry

CENG101 – COMMUNITY ENGAGEMENT

CONTACT TIME (Periods per week)

Theory (3); small group activity & independent study

ASSESSMENT

Continuous Assessment

Reflective Journal (individual) 40%

Creative presentation (group) 10%

Practical project (group) 50%

SYLLABUS: Introduction to the Applied Sciences, Community Engagement and Community Development

This topic includes the multidisciplinary and multi-sectoral nature of community development and engagement; integration in line with the humanistic philosophy and ecosystems theory

Conceptual Framework

Conceptualization of CE in its various forms; DUT conceptualisation of CE; purposes of and rationale for CE in the South African context (including the NDP); active and participatory citizenry; service and social action; social Negotiation of ground rules; Developing of questions and goals for learning; Existing experience of leadership; Basic concepts and theories justice, change and transformation; students as change agents.

Ethics, principles and values for CE (Rules of and for Engagement)

This section includes: humanistic philosophy and Ubuntu; ethics and ethical conduct in relationship with communities; values that guide CE practice in the South African context; principles that guide CE practice in the South African context.

The Approaches and Process of CE

This topic includes dimensions of community (social, political, economic, cultural, physical) and community issues; reciprocity and partnerships; the integrated and holistic approach to education and life-relating discipline/faculty based knowledge to real issues; stages (phases) of the CE process.

Skills and Attributes for CE

This section includes critical thinking; problem solving; communication skills (written and verbal-interviewing, active listening); interaction and human relationships; discipline specific skills; attributes (from character, love for self and humanity, creativity, curiosity for knowledge - discipline/faculty and self)

Challenges and Benefits of CE

Covers Community, institution, faculty and students perspectives.

EXLN201- EXPERIENTIAL LEARNING II (WIL)

ASSESSMENT

Continuous Assessment

Laboratory Work:	50%
Performance Appraisal:	50%

CHPJ201- CHEMISTRY PROJECT II (WIL) ASSESSMENT

Continuous Assessment

Written Report:	50%
Oral Presentation:	50%

12.2 NATIONAL DIPLOMA: ANALYTICAL CHEMISTRY (NDACH2)

Unless otherwise specified, where practicals are constituted as part of the course mark, a sub-minimum of 40% is applicable to the practical mark.

CHEM102 - CHEMISTRY I (150417112)

CONTACT TIME (Periods per week)

Theory (4); Tutorial (1); Practical (3)

ASSESSMENT

Course Mark

Theory: 2 x 1 hour tests (30% each):	Total 60%
Practicals: 2 x 3 hour tests (15% each):	Total 30%
Computer Tutorials and Practical reports:	Total 10%

Final

Mark

Examination: 1 x 3 hour paper

Final mark: Course mark (40%) + Examination mark (60%)

SYLLABUS: Introduction: matter, measurement and molecules; stoichiometry, aqueous reactions and solution stoichiometry; acid-base equilibria; electronic structure of atoms; periodic properties of elements; chemical bonding; organic chemistry nomenclature and properties: alkanes and cycloalkanes; unsaturated hydrocarbons; alcohols, phenols & ethers; aldehydes and ketones; carboxylic acids; and esters; amines and amides.

CSK1103 - COMMUNICATION SKILLS I (129900612)

CONTACT TIME (Periods per week)

Theory (2); Tutorial (1)

ASSESSMENT

Continuous Assessment

1 x 1 hour Theory test:	33.3%
1 x Oral Presentation:	33.3%
1 x Written Project:	33.3%

SYLLABUS: Communication theory; oral presentation; technical writing skills; group communication skills.

CSCC101 - COMPUTER SKILLS I (60205512)

CONTACT TIME (Periods per week)

Theory and demonstration of practical (2); Computer laboratory practical session (2).

ASSESSMENT

Continuous Assessment

Theory Test:	25%
Practical evaluation presentations (or a test):	25%
Word processing:	25%
Spreadsheets:	25%

The practical tests will be set in the students' practical time. (Each group will have their own test). A make-up theory test will be set at the end of the semester. This would be for students who missed the theory test.

SYLLABUS: Computer hardware; software; computer utilization; Practical work will use MS Office.

MATH101 - MATHEMATICS I (160404012)

CONTACT TIME (Periods per week)

Theory (4); Tutorials (2)

ASSESSMENT

Continuous Assessment

Best 3 out of 4 short (30 minute) tests and/or assignments: 20% 2x 1.5 hour Major tests (moderated). One for each section: 80% Pass mark = 50%, with a sub-minimum of 40% for each major test.

Note:

- 13.1.1 Students who obtain a final result between 45% and 49% will be eligible to write a 3 hour make-up test covering the whole syllabus, at the end of the semester, with the students who missed a control (major) test with a valid reason which, if passed, will result in a mark of 50% being allocated.
- 13.1.2 Students who get at least 50% but who fail due to the test sub-minimum also qualify for this test.
- 13.1.3 Students who miss a major test & are allowed to write the make-up test on the work covered in the missed test, DO NOT qualify for the above.

SYLLABUS: Basic mathematics - revision of school work; algebra; trigonometry; statistics calculus; integration.

PHSA102 - PHYSICS I (150710512)

CONTACT TIME (Periods per week)

Theory (3); Tutorial (1); Practical (3).

ASSESSMENT

Course Mark

Theory: Best 2 out of 3 theory tests (1 hour each):	65%
Practicals: Practicals (15%) & Test (85%):	30%
Tutorial Tests:	5%

Final Mark

Examination: 1 X 3 Hour theory examination

Final mark: Course Mark (40%) + Examination Mark (60%)

SYLLABUS: Introduction to physics; vectors and scalars; motion with constant acceleration; newton's laws of motion; application of newton's laws; work, energy and power; elasticity, stress and strain; static fluids; temperature measurement; thermal expansion; heat and calorimetry; electrostatics; direct current circuits; magnetism; wave motion; general optics; structure of the atom and the nucleus; radioactivity.

ACHM103 - ANALYTICAL CHEMISTRY I (150413512)

Prerequisite: CHEM102

CONTACT TIME (Periods per week)

Theory (4); Tutorials (1); Practical (6).

ASSESSMENT

Course Mark

Theory: 2 x 1 hour Tests (20% each):	40%
Assignment:	10%
Practical: 1 x 1 hour Theory of Practical Test:	15%
1 x 3 hour Practical test:	15%
Practical write-ups:	20%

Final Mark

Examination: 1 x 3 Hour Theory Examination

Final mark: Course Mark (40%) + Examination Mark (60%)

SYLLABUS: Introduction; elementary statistics; laboratory practice and safety; titrimetric analysis; sampling and sample handling; redox titrations; introduction to instrumental analysis; basic report writing

INCH201 - INORGANIC CHEMISTRY II (150416822)

Prerequisite: CHEM102

CONTACT TIME (Periods per week)

Theory (2); Tutorial (1); Practical (2).

ASSESSMENT

Course Mark

Theory: 2 x 1 hour tests - 25% each:	50%
Theory of Practical:	10%
Practicals:	40%

Final Mark

Examination: 1 x 3 hour theory paper

Final mark: Course Mark (40%) + Examination Mark (60%)

SYLLABUS: Chemical bonding; solution chemistry; descriptive chemistry of hydrogen and selected elements in Groups 1, 2, 13, 14, 15, 16 and 17.

ORCH201 - ORGANIC CHEMISTRY II (150413822)

Prerequisite: CHEM102

CONTACT TIME (Periods per week)

Theory (2); Tutorial (1); Practical (2).

ASSESSMENT

Course Mark

Theory: 2 x 1 hour tests - 25% each: 50%

Assignment: 10%

Practical mark: 40%

Final Mark

Examination: 1 x 3 hour paper

Final mark: Course Mark (40%) + Examination Mark (60%)

SYLLABUS: Acids and bases in organic chemistry; aliphatic hydrocarbons; radical reactions; ionic reactions; alcohols and ethers; aromatic compounds; electrophilic and nucleophilic substitution; aldehydes and ketones; carboxylic acids and derivatives; amines and amides.

PHCH201 - PHYSICAL CHEMISTRY II (150413722)

Prerequisite: CHEM102

Corequisite: MATH101

CONTACT TIME (Periods per week)

Theory (2); Tutorial (1); Practical (3).

ASSESSMENT

Course

Mark

Theory: 2 x 1 Hour tests - 25% each: 50%

Theory of Practical: 10%

Practical Reports: 40%

Final Mark

Examination: 1 x 3 hour paper

Final Mark: Course Mark (40%) + Examination Mark (60%)

SYLLABUS: Gases; liquids; chemical equilibrium; colligative properties of solutions; electrochemistry; reaction kinetics; colloids.

ACHM204 - ANALYTICAL CHEMISTRY II (150416622)

Corequisite: ACPR201

CONTACT TIME (Periods per week)

Theory (4); Tutorial (1).

ASSESSMENT

Course Mark

Theory: 2 x 1 hour theory tests: 90%

Assignments: 10%

Final Mark

Examination: 1 x 3 hour theory paper

Final Mark: Course Mark (40%) + Examination mark (60%)

SYLLABUS: Principles of analytical chemistry; introduction; basic statistics; sampling and sample handling; methods of analysis; gravimetric analysis; titrimetric analysis; instrumental techniques.

ACPR201 - ANALYTICAL CHEMISTRY PRACTICAL: II (150400622)

CONTACT TIME (Periods per week)

Practical (9).

ASSESSMENT

Continuous Assessment

1) Wet Chemistry Practicals:

2 Practical Tests (including theory of prac sections) 40%

Continuous Assessment (pre-labs and prac reports): 25%

Total (Wet Chemistry): 65 %

2) Instrumental Practicals:

1 Theory of Practical Test: 20%

Continuous Assessment (pre-labs and prac reports): 15%

Total (Instrumental Analysis): 35 %

SYLLABUS: Wet chemistry; gravimetric analysis; titrimetric analysis. Instrumental analysis;

ORCH302 - ORGANIC CHEMISTRY III (150414103)

CONTACT TIME (Periods per week)

Theory (2); Practical (3).

ASSESSMENT (per module)

Course Mark

Theory: 2 x 1 hour tests (25 % each): 50%

Theory of Practical: 10%

Practical mark: 40%

Final Mark

Examination: 2 x 1½ hour papers

Final mark: Course mark (40%) + Examination Mark (60%)

SYLLABUS:

Paper 1 - acids and bases in organic chemistry; conformation of molecules;

stereochemistry; elimination and substitution reactions; polymers; carbenes; aromatic compounds; aldehydes and ketones; dicarbonyl compounds.

Paper 2; lipids; amino acids and proteins; spectroscopy.

PHCH301 - PHYSICAL CHEMISTRY III (150414303)

CONTACT TIME (Periods per week)

Theory (2); Practical (3).

ASSESSMENT (per module)

Course Mark

Theory: 2 theory tests: 50%

Assignment: 10%

Practical mark: 40%

Final Mark

Examination: 1 x 3 hour theory paper

Final Mark: Course mark (40%) + Examination Mark (60%)

SYLLABUS: Paper 1; electrochemistry; solid state.

Paper 2 - Quantum mechanics and spectroscopy; changes of phase; kinetics; surface Chemistry.

INCH301 - INORGANIC CHEMISTRY III (150414203)

CONTACT TIME (Periods per week)

Theory (2); Practical (3).

ASSESSMENT (per module)

Course Mark

Theory: 2 theory tests: 50%

Assignment: 10%

Practical mark: 40%

Final Mark

Examination: 2 X 1½ hour theory papers

Final mark: Course Mark 40% + Examination Mark 60%

SYLLABUS: Paper 1 - Theories of Bonding; Co-ordination Chemistry and Crystal Field Theory;

Paper 2 - Descriptive Chemistry of 1st Transition Series, Associated Elements and Nuclear Chemistry

CQAS201 - CHEMICAL QUALITY ASSURANCE (150416722)

Prerequisite: ACHM103

CONTACT TIME (Periods per week)

Theory (4); Tutorial (1).

ASSESSMENT

Course Mark

Theory: Tests: 2 x 1 hour tests (weighted 40% each): 80%

Assignment: 20%

Final Mark

Examination: 1 x 3 hour theory paper

Final Mark: Course Mark (40%) + Examination mark (60%)

SYLLABUS: Quality Assurance Systems; Laboratory Accreditation; Advanced Statistical Treatment of Data in Analytical Chemistry; Sample preparation.

ACHM303 - ANALYTICAL CHEMISTRY III comprises ACHM313 and ACHM323 in a 50%:50% weighting.**ACHM313 - ANALYTICAL CHEMISTRY III MODULE I**

Prerequisite: ACHM204 **Co-requisite:** ACPR302.

CONTACT TIME (Periods per week)

Theory (4).

ASSESSMENT**Course Mark**

Theory: 2 x 1 hour tests: 85%

Assignment / Tutorials: 15%

Final Mark

Examinations: 2 x 1 ½ hour theory examinations

Final Mark: Course Mark (40%) + Examination Mark (60%)

SYLLABUS: Atomic spectroscopy; molecular spectroscopy.

ACHM323 - ANALYTICAL CHEMISTRY III MODULE 2

Prerequisite: ACHM204 **Corequisite:** ACPR302.

CONTACT TIME (Periods per week) Theory (4).

ASSESSMENT**Course Mark**

Theory: 2 x 1 hour tests: 85%

Assignment / Tutorials: 15%

Final Mark

Examinations: 2 x 1 ½ hour theory examinations

Final Mark: Course Mark (40%) + Examination Mark (60%)

SYLLABUS: Chromatographic methods; electro-analytical techniques; thermal analysis.

ACPR302 - ANALYTICAL CHEMISTRY: PRACTICAL III comprises

ACPR312 and ACPR322 in a 50%:50% weighting.

ACPR312 - ANALYTICAL CHEMISTRY: PRACTICAL III MODULE I

CONTACT TIME (Periods per week)**Practical (6)****ASSESSMENT****Continuous Assessment**

Comprehensive Reports:	40%
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Practical Tests:	50%
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Project:	10%
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SYLLABUS: Atomic spectroscopy; molecular spectroscopy.**ACPR322 - ANALYTICAL CHEMISTRY: PRACTICAL III
MODULE 2****CONTACT TIME (Periods per week)****Practical (6)****ASSESSMENT****Continuous Assessment**

Comprehensive Reports:	40%
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Practical Tests:	50%
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Project:	10%
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SYLLABUS: Chromatographic methods; electroanalytical techniques; thermal analysis.**WORK-INTEGRATED LEARNING EXAM101 LEARNING****Prerequisites:** ACHM313, ACHM323, ACPR312 and ACPR322 **DURATION:**

6 months in industry

ASSESSMENT

Laboratory Work:	50%
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Written Report:	30%
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Performance Appraisal:	20%
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CMPJ301 PROJECT III**Prerequisite:** EXAM101**DURATION:** 6 months in industry or 8 hours per week**ASSESSMENT**

Written Report:	50%
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Oral Presentation:	25%
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Poster Presentation:	25%
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12.3 ADVANCED DIPLOMA IN CHEMISTRY

ANAC401 – ANALYTICAL CHEMISTRY IV

CONTACT TIME (Periods per week)

Theory and demonstration of practical (11)

ASSESSMENT

Continuous Assessment

Theory Test:	60%
Practical evaluation/test:	30 %
Assignment, portfolios, presentations:	10%

SYLLABUS: ICP-VGA/ETA, X-Rays, ICP-MS, NIR/MIR, Raman, Pulse Methods and CV, Thermal Methods, LC/GC-MS

ORGC401-ORGANIC CHEMISTRY IV

CONTACT TIME (Periods per week)

Theory and demonstration of practical (7)

ASSESSMENT

Continuous Assessment

Theory Test:	60%
Practical evaluation/test:	30 %
Assignment, portfolios, presentations:	10%

SYLLABUS: Infra-Red, Mass Spectroscopy and Nuclear Magnetic Resonance Spectroscopy, Alkaloids, Terpenes, Pericyclic Chemistry, Enolates and Retrosynthesis

INOC401- INORGANIC CHEMISTRY IV

CONTACT TIME (Periods per week)

Theory and demonstration of practical (7)

ASSESSMENT

Continuous Assessment

Theory Test:	60%
Practical evaluation/test:	30 %
Assignment, portfolios, presentations:	10%

SYLLABUS: Catalysts Heterogeneous and Homogenous Catalysts: Ziegler-Natta and Vanadium pentoxide; Organometallics; Metallurgy: Platinum group metals, First Transition series metals and Coinage metals

PHYC401- PHYSICAL CHEMISTRY IV

CONTACT TIME (Periods per week)

Theory and demonstration of practical (7)

ASSESSMENT

Continuous Assessment

Theory Test:	60%
Practical evaluation/test:	30 %
Assignment, portfolios, presentations:	10%

SYLLABUS: Advanced thermodynamics: Entropy and Free Energy, Thermochemistry; Advanced electrochemistry: Faradaic Processes, Electrochemical Equations, Liquid-junction Potentials; Advanced chemical kinetics: Reaction Rates, Complex Reactions, Enzyme Catalysis; Advanced surface chemistry: Surface Tension: Young and Laplace Equations; Spreading Dynamics and Adsorption Isotherms

CHEP401- CHEMISTRY PROJECT IV

CONTACT TIME (Periods per week)

Theory (2/w term 2)

ASSESSMENT

Continuous Assessment

Research Proposal: written report and oral presentation 30%

Research Project: written report and oral presentation 70%

SYLLABUS: Guidelines for research proposals and preparation of a research proposal, literature review, critical evaluation of literature papers, project management skills in chemistry, laboratory skills, analyses of results, problem solving in relation to the project, outline of industry standards, particularly within the laboratory, Mathematical and Information Technology skills used in the operation of statistical packages, writing of a research report based on the research project, oral presentation of the research project and Harvard method for referencing.

12.4 POSTGRADUATE DIPLOMA IN CHEMISTRY

Unless otherwise specified, where practicals are constituted as part of the course mark, a sub-minimum of 40% is applicable to the practical mark.

ANCH801 – ANALYTICAL CHEMISTRY V

CONTACT TIME (Periods per week)

Theory (3). Practical (3)

ASSESSMENT

Course Mark

Theory: Minimum 2 x tests	60%
Practical:	40%

Final Mark

Examinations: 1 x 3 hour theory examination

Final Mark: Course Mark (40%) + Examination Mark (60%)

SYLLABUS: Atomic Spectroscopy, Chromatography, Thermal Methods, Kinetic Methods, Electrochemistry, Molecular Spectroscopy and Microscopic techniques.

CHRP802 - CHEMISTRY PROJECT V

ASSESSMENT

CONTINUOUS ASSESSMENT

Course Mark

Oral Presentation	40%
Written Report	60%

Final Mark

Continuous assessment: Minimum mark of 50% is required to pass this module.

SYLLABUS: The following are to be done in conjunction with the supervisor:

1. Research methodology
- 1.1 Reference styles and tools
- 1.2 Statistical packages
- 1.3 Plagiarism – to run the Turnitin program
2. Critical evaluation of literature papers
3. Writing of a research report based on the research project
4. Oral presentation of the research project

INCH801 – INORGANIC CHEMISTRY V

CONTACT TIME (Periods per week)

Theory (3) Practical (3)

ASSESSMENT

Course Mark

Theory: Minimum 2 x tests	60%
Practical:	40%

Final Mark

Examinations: 1 x 3 hour theory examination

Final Mark: Course Mark (40%) + Examination Mark (60%)

SYLLABUS: Second and third row transition elements; Chemistry of Lanthanides and Actinides; Spectroscopy: Tanabe-Sugano diagrams, Symmetry groups and infrared; Chemistry of materials; Green chemistry; Techniques of characterization; Nanochemistry and nanotechnology; Synthesis of organometallic-ionic liquid catalysts.

ORCH801 - ORGANIC CHEMISTRY V

CONTACT TIME (Periods per week)

Theory (3). Practical (3)

ASSESSMENT

Course Mark

Theory: Minimum 2 x tests 60%

Practical: 40%

Final Mark

Examinations: 1 x 3 hour theory examination

Final Mark: Course Mark (40%) + Examination Mark (60%)

SYLLABUS: Coupling reactions; Multi-component reactions; Heterocyclic chemistry; Stereochemistry; Green Chemistry; Retrosynthesis of poly-functional molecules; 2 D-NMR spectroscopy (COSY, HMBC, NOESY, HSQC, DEPT, TOCSY); Poly-functional organic molecules by MS.

PHCH802 – PHYSICAL CHEMISTRY V

CONTACT TIME (Periods per week)

Theory (3). Practical (3)

ASSESSMENT

Course Mark

Theory: Minimum 2 x tests 60%

Practical: 40%

Final Mark

Examinations: 1 x 3 hour theory examination

Final Mark: Course Mark (40%) + Examination Mark (60%)

SYLLABUS: Advanced phase separation technology; Introduction to computational theory and calculations in the application of advanced thermodynamics; Electrochemical methods to understand, evaluate and solve corrosion chemistry problems; Electroplating theory and techniques; Fractionation and application of biomass components; Nanotechnology based electrochemical application; Photovoltaic cells theory and applications; Solid state chemistry and crystallography; Development of biosensors.

SCGY801 - SCIENTIFIC METHODOLOGY

CONTACT TIME (Periods per week)

Theory (1)

ASSESSMENT

Course Mark

One Assignment (literature review)	20%
1 x Test (Statistics)	20%
1 x Proposal writing skills evaluation (proposal report)	20%
1 x Oral presentation (proposal)	40%

Final Mark

A final mark of 50% is required to pass the module

SYLLABUS: The nature of scientific methodology; Formulating a research topic; Scientific research strategy. The use of statistics for experimental and data analysis design. Guidelines for research proposals and preparation of a research proposal. Literature review stating the aims and significance of the proposed research in conjunction with relevant literature papers.

E&OE